

ACTIVATION BY Hg^{2+} OF ACETOACETYL-CoA REDUCTASE IN EXTRACTS OF *RHODOPSEUDOMONAS SPHEROIDES* AND *RHODOMICROBIUM VANNIELII*

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1. Introduction

In examining the activities of enzymes concerned in poly- β -hydroxybutyrate formation in *Athiorhodaceae* an activation of acetoacetyl-CoA reductase (EC 1.1.1.36, D-3-hydroxyacetyl-CoA: NAD(P) oxidoreductase) by Hg^{2+} was observed. Some properties of this unusual activation are examined in cell-free extracts of several bacterial species and in a partially purified preparation of acetoacetyl-CoA reductase from *Rhodopseudomonas spheroides*.

2. Methods

The non-sulphur, purple bacteria, *Rhodopseudomonas spheroides* (N.C.I.B. No. 8253) and *Rhodomicrobium vannielii*, were grown photosynthetically as described earlier [1] on a malate-glutamate medium [2]. Other bacteria were grown on nutrient broth at 30° in an Orbital Incubated Shaker. Organisms were harvested in late log phase of growth, washed once with 10 mM sodium phosphate, pH 7.5, and resuspended in a small volume of the same buffer. After ultrasonic treatment at 0° for three minutes unbroken cells, debris and some particulate material were removed by centrifugation at 40,000 g for 2 hr at 0°. The supernatant fraction from this used for enzyme assays and protein concentration was determined by the procedure of Layne [3].

Acetoacetyl-CoA reductase was assayed by the measurement of ΔE_{340} due to the oxidation of NADH in the presence of the artificial substrate *S*-acetoacetyl-*N*-acetylcysteamine, this being preferable to the

natural substrate acetoacetyl-CoA [4]. Assays were carried out using a Unicam SP700 recording spectrophotometer, at room temperature. Cuvettes contained, in μmole , sodium pyrophosphate-HCl buffer, pH 7.5, 200; NADH, 0.5; *S*-acetoacetyl-*N*-acetylcysteamine, 10 and bacterial extract 0.5-3.0 mg in a total volume of 3 ml. Specific activity of enzyme activity was expressed as μmole NADH oxidised/min/mg protein.

S-Acetoacetyl-*N*-acetylcysteamine was prepared by slow addition of diketone (3.94 g) in ether to an ice cool solution of *N*-acetylcysteamine (5.7 g) in ether. After standing at room temperature for 6 hr the oily mixture was chilled to -10° and the crystalline product collected by filtration. The *N*-acetylcysteamine was prepared by reaction under reflux of thioacetic acid with ethylenimine [5].

3. Results and Discussion

The optimum pH of acetoacetyl-CoA reductase, in extracts of *R. vannielii*, was 7.4 and the effect of addition of metal ions to the standard assay mixture is shown in table 1. Relatively little effect was observed except by Hg^{2+} and, to a lesser extent, by Cd^{2+} . The increase in activity was seven fold in extracts of *R. vannielii* and three fold in extracts of *R. spheroides* (fig. 1). Dialysis of extracts against sodium pyrophosphate buffer increased slightly the activation by Hg^{2+} and this stimulation by Hg^{2+} was not removed by the addition of EDTA (mM) to the assay system (fig. 1). *p*-Chloromercuribenzoate (0.2 mM) produced a similar activation to that of Hg^{2+} in extracts from three other species of *Athiorhodaceae* (*Rhodopseudomonas*

Table 1

Effect of metal ion on the acetoacetyl-CoA reductase activity in *Rhodopseudomonas vannielii*.

| Addition | Concentration (mM) | Alteration in acetoacetyl-CoA reductase activity, relative to control |
|------------------|--------------------|-----------------------------------------------------------------------|
| None | — | 100 |
| Hg ²⁺ | 0.10 | 200 |
| Hg ²⁺ | 0.20 | 700 |
| Hg ²⁺ | 0.30 | 350 |
| Mn ²⁺ | 0.05 | 150 |
| Mn ²⁺ | 0.20 | 100 |
| Cd ²⁺ | 0.30 | 200 |
| Na ²⁺ | 0.20 | 100 |
| AsO ₄ | 0.50 | 80 |
| Pb ²⁺ | 0.50 | 80 |

Enzymic activity was measured as described in Methods, except as indicated in the table; metal ions added last and results are expressed relative to a control.

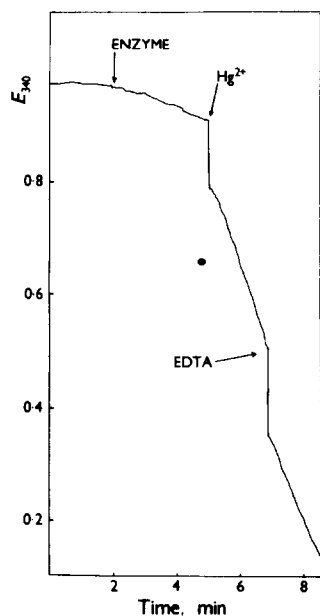


Fig. 1. Activation of acetoacetyl-CoA reductase from *Rhodococcus ruber* by the addition of 0.2 mM Hg²⁺, and the lack of reversal by EDTA (mM).

capsulata, *Rhodopseudomonas gelatinosa*, *Rhodopseudomonas palustris*) or from four species of aerobic heterotrophic bacteria was found.

The complete reaction mixture was necessary, prior to the addition of Hg²⁺, for activation to occur. When Hg²⁺ was preincubated with the enzyme, in the absence of substrate some degree (approx. 10%) of inhibition occurred. Gel filtration on Sephadex G-200 produced a fraction purified 15 fold from *R. spheroides*, which exhibited the same activation by Hg²⁺ as obtained in the intact cell free extract. Variation in substrate concentration in the presence and absence of Hg²⁺ produced significantly different results. The inclusion of Hg²⁺ in the reaction mixture caused the substrate concentration-enzymic rate curve to become more sigmoidal (fig. 2).

The non-enzymic interaction of Hg²⁺ with reduced pyridine nucleotides, and the consequent cautions necessary in interpreting the effect of Hg²⁺ on enzyme assays, has been recently reported [6]. It was suggested that Hg²⁺ forms a weak bond with NADH or NADPH, but that in the presence of a chelating agent, or protein, no such association occurred. The activation of malate dehydrogenase from pig heart mitochondria by Hg²⁺ has been described by Kuramitsu [7] which exhibits similarities to the activation by Hg²⁺ of acetoacetyl-CoA reductase reported here. In both cases it is necessary for the substrate to be present during activation of the enzyme by Hg²⁺; it would appear that a conformational change occurs, caused perhaps by the breakage of SH-group linkages,

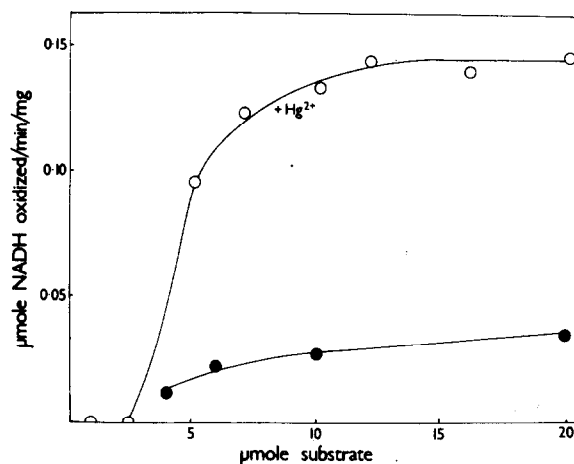


Fig. 2. Acetoacetyl-CoA reductase in extracts of *Rhodopseudomonas spheroides* with substrate (*S*-acetoacetyl-*N*-acetylcysteamine) concentration in the presence and absence of Hg²⁺ (0.2 mM).

that results in activation only when the catalytic site is occupied. Whilst clearly this is not a physiological mode of activation within the bacterial cell it may be that Hg^{2+} mimics the effect of a natural activator.

Acknowledgements

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Reference

- [1] N.G.Carr, and G.Exell, *Biochem. J.* 96 (1965) 688.
- [2] J.Lascelles, *Biochem. J.* 72 (1959) 508.
- [3] E.Layne, *Methods in Enzymology* 3 (1957) 447.
- [4] F.Lynen and O.Wieland, *Methods in Enzymology* 1 (1955) 566.
- [5] R.Kuhn and G.Quadbeck, *Chem. Ber.* 84 (1951) 844.
- [6] J.D.McGarry, P.Donkin and H.K.King, *Arch. Biochem. Biophys.* 126 (1968) 973.
- [7] H.K.Kuramitsu, *J. Biol. Chem.* 243 (1968) 1016.